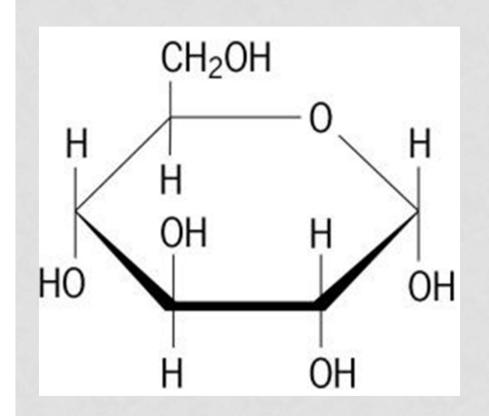
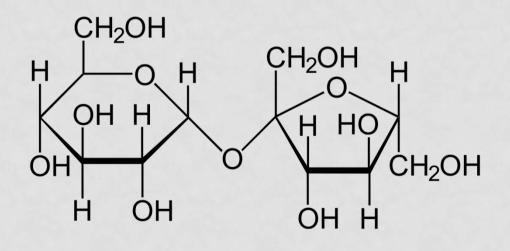
- Compounds made of C, H, O
 - Ratio of 1 C: 2 H: 1 O
 - Example: $C_6H_{12}O_6$
- C,H,O can connect over and over to make a long chain
- Smallest unit (monomer) = monosaccharide
 - Can also be called simple sugars
 - Usually have 3 to 7 units in the compound
 - Example: Glucose

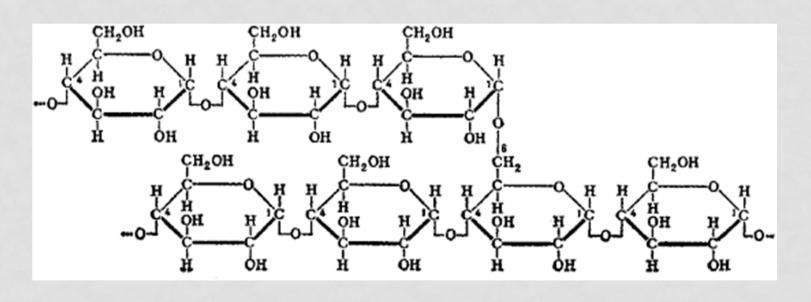


- How many
 - C
 - H
 - ()
- What type of structure?
- <u>Glucose</u> energy source for organisms

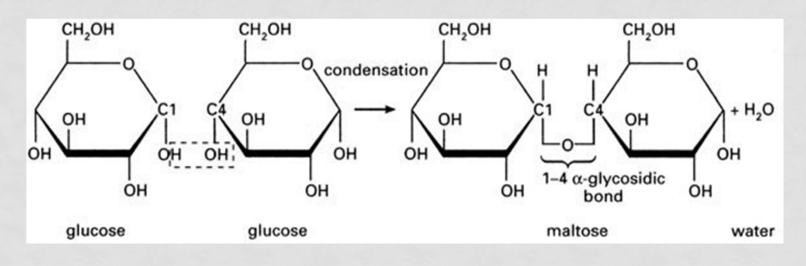
- Monosaccharides combine to make other compounds
 - <u>Disaccharide</u> 2 monosaccharides linked together to form energy sources
 - Example: Sucrose (table sugar)



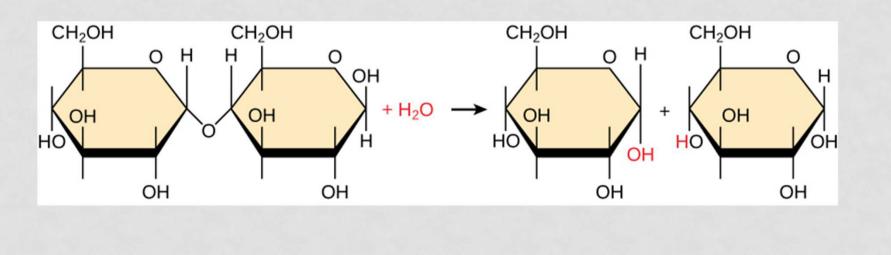
- More monosaccharides can make longer compounds
 - <u>Polysaccharide</u> more than one monosaccharide linked together
 - Example: Glycogen (energy source in liver)



- To connect monosaccharides together, a chemical reaction has to take place
- <u>Condensation</u> bonds two monomers together
 - makes water
 - One monomer loses an H, the other loses O and H
 - Together they form H₂O



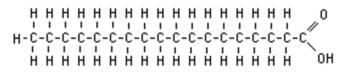
- Your body has to be able to break down carbohydrates as well
- <u>Hydrolysis</u> the addition of water breaks apart a bond
 - add in 2 H, and 1 O to break the bond between monomers
 - "hydro" = water "lysis" = cut



- Roles of carbohydrates
 - Short term energy sources
 - Structural support in plant cell walls
 - Hard outer shell of insects, shrimp and lobsters

LIPIDS

- Made mostly of C and H
- Purpose store energy
- Make up fats, oils and waxes
- Made of two parts:
 - Fatty acids
 - Glycerol



long hydrocarbon chain carboxylic acid group

Essential features of a fatty acid

