DIFFUSION

 Diffusion can take place at different rates (speeds)

Concentration

- High concentration = faster diffusion
- More particles to that want to move

• <u>Temperature</u>

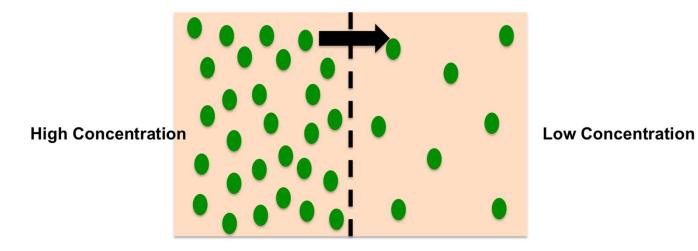
- High temperature = faster diffusion
- More particles collide which makes them more

• Pressure

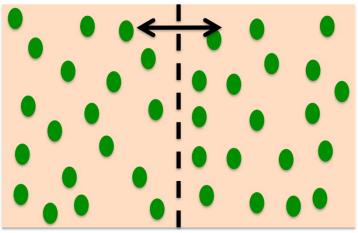
- High pressure = faster diffusion
- Particles closer together which make more collisions



DIFFUSION



Balanced Concentration





DIFFUSION ACROSS THE PLASMA MEMBRANE

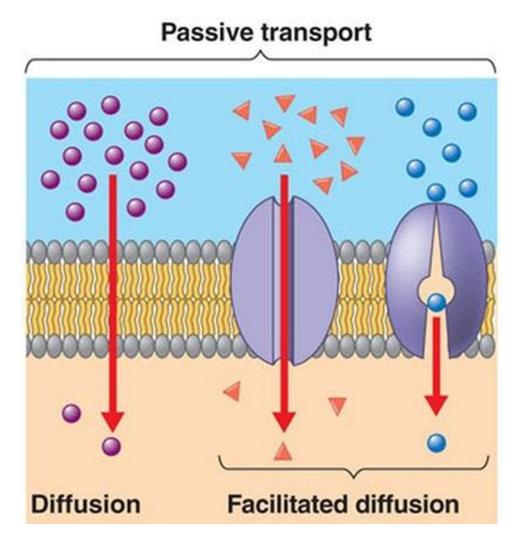
- Cells need water AND other molecules to get in and out of the cell
- Water diffuses through the plasma membrane very easily
- Most other molecules CANNOT move as easily

• Facilitated diffusion

- Uses transport proteins to move small molecules across the membrane
- Still requires NO EXTRA ENERGY
- Still moving from high concentration to low concentration



DIFFUSION ACROSS THE PLASMA MEMBRANE





OSMOSIS: DIFFUSION OF WATER

- Water passes freely into and out of the cell
- Osmosis diffusion of water across a selectively permeable membrane
- Need to regulate water movement to maintain homeostasis
- Demonstration:
 - Solution of cornstarch inside the tubing
 - Placed in a beaker of iodine
 - What will happen over the hour?



HOW OSMOSIS WORKS

- Vocabulary review:
 - Solution
 - Solvent
 - Solute
- Water is the solvent in a cell
- Concentration tells how much solute is dissolved in a solvent
 - Remember it's talking about particles
- When solvent increases, concentration decreases
- When solvent decreases, concentration increases



HOW OSMOSIS WORKS

